

Central problem: How to calculate macroscopic, time-averaged properties from rapidly fluctuating microscopic quantities?

Brute force approach: Time-average over the microscopic properties

$f_{obs}$   $\equiv$  observed macroscopic property - pressure, etc.

$f(\underline{\mathbf{q}}^{3N}, \underline{\mathbf{p}}^{3N})$   $\equiv$  microscopic mechanical variable

$$f_{obs} = \lim_{\tau \rightarrow \infty} \frac{1}{\tau} \int_0^\tau f(\underline{\mathbf{q}}^{3N}, \underline{\mathbf{p}}^{3N}) d\tau' \quad \text{Time average}$$

But this requires calculation of time-dependent trajectories for all  $N$  particles!

Better approach: ENSEMBLE THEORY

Developed by J. Willard Gibbs - founder of statistical mechanics

Replaces time average with ensemble average

**Ensemble**  $\equiv$  collection of all possible states of an assembly

e.g. assembly of only 2 particles quantum description

Constant energy ensemble with 7 quanta of translational energy

State	$n_{1x}$	$n_{1y}$	$n_{1z}$	$n_{2x}$	$n_{2y}$	$n_{2z}$
$\alpha$	2	1	1	1	1	1
$\beta$	1	2	1	1	1	1
$\gamma$	1	1	2	1	1	1
$\delta$	1	1	1	2	1	1
$\epsilon$	1	1	1	1	2	1
$\eta$	1	1	1	1	1	2

These are all of the 7-quanta states

$$E_\alpha = \sum_{i=1}^{N=2} \epsilon_i$$

$$\epsilon_i = \frac{h^2}{8ma^2} (n_{ix}^2 + n_{iy}^2 + n_{iz}^2) = \frac{h^2}{8ma^2} (3 \text{ or } 6)$$

classical description

specify all the position & momentum variables

$p_{1x} p_{1y} p_{1z} p_{2x} p_{2y} p_{2z}$

$q_{1x} q_{1y} q_{1z} q_{2x} q_{2y} q_{2z}$

$$E_\alpha = \sum_{i=1}^{N=2} \epsilon_i \quad \epsilon_i = (p_{ix}^2 + p_{iy}^2 + p_{iz}^2) / 2m$$

The  $p$  values squared must add to give the correct total energy.

QM ensemble average is a sum over states  
 CM ensemble average is an integral over states

**ERGODIC HYPOTHESIS: Time average  $\Leftrightarrow$  Ensemble average**

**Ensemble average for macroscopic property  $f$**

$$\bar{f} = \sum_j P_j f_j \quad P_j \equiv \text{probability that assembly is in distinguishable assembly state } j$$

$$\sum_j P_j = 1 \quad \text{probabilities are normalized}$$

e.g. ensemble average energy:  $\bar{E} = \sum_j P_j E_j$  note  $E_j$  are assembly energies

**Ensemble average for continuous variables - classical treatment**

$$\bar{f} = \int \cdots \int d\mathbf{q}^{3N} d\mathbf{p}^{3N} P(\mathbf{q}^{3N}, \mathbf{p}^{3N}) f(\mathbf{q}^{3N}, \mathbf{p}^{3N})$$

where  $P(\mathbf{q}^{3N}, \mathbf{p}^{3N}) d\mathbf{q}^{3N} d\mathbf{p}^{3N} \equiv$  probability of assembly being in volume element  $d\mathbf{q}^{3N} d\mathbf{p}^{3N}$  centered at  $(\mathbf{q}^{3N}, \mathbf{p}^{3N})$

In either QM or CM case, we need a complete list of all the *distinguishable* assembly states and their probabilities  $P_j$ . How do we determine  $P_j$ ?

They must give the minimum free energy under the experimental conditions!  
 e.g. minimum Helmholtz free energy  $A$  if we have fixed  $(N, V, T)$

CANONICAL ENSEMBLE  $\equiv$  subject to constraint of constant  $(N, V, T)$   
 - closed, thermodynamically stable system

The states of the assembly, given by  $\{P_j\}$ , must minimize  $A$ .

We need to write  $A$  in terms of the  $P_j$  values. How?

$$A = E - TS \quad \bar{E} = \sum_j P_j E_j \quad \Rightarrow \quad A = \sum_j P_j E_j - TS$$

What about entropy  $S$ ? The connection between  $S$  and  $\{P_j\}$  is *assumed* to be.....

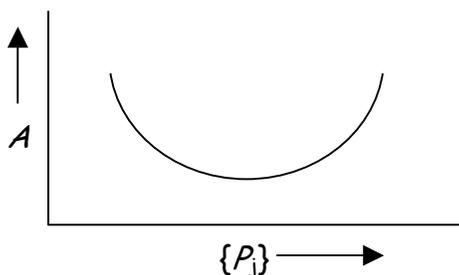
$$S = -k \sum_j P_j \ln P_j$$

Central assumption of Boltzmann (originally in somewhat different form that we'll see shortly). No derivation - only plausibility arguments. Statistical mechanics is built on this assumption!

$k = R/N_A = 1.38 \times 10^{-23}$  J/K  $\equiv$  Boltzmann constant

$$A = E - TS = \sum_j P_j E_j + kT \sum_j P_j \ln P_j = \sum_j P_j (E_j + kT \ln P_j)$$

To find the  $\{P_j\}$  values that minimize  $A$ , imagine the real assembly at equilibrium, with the minimum  $A$  and the probabilities  $\{P_j\}$ , and other assemblies with non-equilibrium  $A$  and different  $\{P_j\}$  values.



Let  $\{P_j\} \rightarrow \{P_j + \delta P_j\}$

Then  $A \rightarrow A + \delta A$

Find  $\{\delta P_j\}$  such that  $\delta A = 0$

$$\delta A = \delta \left[ \sum_j P_j (E_j + kT \ln P_j) \right] = \sum_j (E_j \delta P_j) + kT \ln P_j \delta P_j + kT P_j \frac{1}{P_j} \delta P_j = 0$$

$$\sum_j \delta P_j [E_j + kT (\ln P_j + 1)] = 0$$

Introduce constraint  $\sum_j P_j = 1$

After  $P_j \rightarrow P_j + \delta P_j$  still  $\sum_j (P_j + \delta P_j) = 1$

Probabilities still add to 1 before or after the change  $\{\delta P_j\}$ .

$$\text{Then } \sum_j \delta P_j = 0 \Rightarrow \delta P_1 = -\sum_{j=2}^N \delta P_j$$

$$\delta A = \delta P_1 [E_1 + kT(\ln P_1 + 1)] + \sum_{j=2}^N \delta P_j [E_j + kT(\ln P_j + 1)] = 0$$

$$\delta A = \sum_{j=2}^N \delta P_j [(E_j - E_1) + kT(\ln P_j - \ln P_1)] = 0$$

The  $\delta P_j$ 's from  $j = 2$  to  $N$  are completely independent, for arbitrary  $\{\delta P_j\}$ , so

$$(E_j - E_1) + kT(\ln P_j - \ln P_1) = 0 \quad \Rightarrow \quad \frac{P_j}{P_1} = \frac{e^{-E_j/kT}}{e^{-E_1/kT}}$$

$$P_j = P_1 e^{E_1/kT} e^{-E_j/kT} \quad \sum_j P_j = 1 = P_1 e^{E_1/kT} \sum_j e^{-E_j/kT}$$

$$P_1 = \frac{e^{-E_1/kT}}{\sum_j e^{-E_j/kT}} \quad \text{same for } P_2, P_3, \dots, P_n, \text{ any } P_j$$

$$P_n = \frac{e^{-E_n/kT}}{\sum_j e^{-E_j/kT}} \quad \text{or} \quad P_j = \frac{e^{-E_j/kT}}{\sum_j e^{-E_j/kT}}$$

Canonical Distribution Function gives the probability for the  $j^{\text{th}}$  distinguishable state in the ensemble. This distribution minimizes  $A \Rightarrow$  *equilibrium distribution*.

We needed the key assumption  $S = -k \sum_j P_j \ln P_j$

This leads to the result that  $P_i$  depends on  $E_i$  only

- $\Rightarrow$  equal energy states have equal probabilities (seems highly plausible)
- $\Rightarrow$  probability decreases exponentially w/ energy (familiar dependence)
- $\Rightarrow$  probability of high-energy state increases with  $T$  (also familiar)

Denominator has special name.....**CANONICAL PARTITION FUNCTION  $Q$**

$$Q(N, V, T) = \sum_j e^{-E_j/kT}$$

Sum of "Boltzmann factors"  $e^{-E_j/kT}$  over all the assembly states  
Originally called "Zustandsumme"  $\equiv Z \equiv$  "sum over states"

$Q$  is a very important quantity! So let's rewrite  $P_j$  in terms of it:

$$P_j = \frac{e^{-E_j/kT}}{\sum_j e^{-E_j/kT}} = \frac{e^{-E_j/kT}}{Q}$$

We'll be able to use  $Q$ , instead of any individual  $P_j$  values, to calculate everything! e.g. calculation of energy  $\bar{E}$ :

$$\bar{E} = \sum_j P_j E_j = f(Q)$$

Define  $\beta = 1/kT$  so differentiation is simpler

$$Q(N, V, T) = \sum_j e^{-E_j/kT} = \sum_j e^{-\beta E_j} \quad \frac{\partial Q}{\partial \beta} = -\sum_j E_j e^{-\beta E_j}$$

$$\text{Recall } P_j = \frac{e^{-E_j/kT}}{Q} \text{ so } e^{-\beta E_j} = Q P_j \Rightarrow \frac{\partial Q}{\partial \beta} = -\sum_j P_j E_j Q = -Q \sum_j P_j E_j = -Q \bar{E}$$

$$\bar{E} = -\frac{1}{Q} \frac{\partial Q}{\partial \beta} = -\frac{\partial \ln Q}{\partial \beta} = -\frac{\partial \ln Q}{\partial (1/kT)} \quad \boxed{\bar{E} = kT^2 \frac{\partial \ln Q}{\partial T}}$$

Ensemble average energy  $\bar{E}$  in terms of  $Q$ , not  $P_j$ .

How about entropy?

$$S = -k \sum_j P_j \ln P_j = -k \sum_j P_j \ln \left( \frac{e^{-E_j/kT}}{Q} \right) = -k \sum_j P_j \left( -\frac{E_j}{kT} - \ln Q \right) = \frac{\sum_j P_j E_j}{T} + k \ln Q$$

$$\boxed{S = k \ln Q + \frac{\bar{E}}{T} = k \ln Q + kT \left( \frac{\partial \ln Q}{\partial T} \right)_{N, V}}$$

Writing all thermodynamic functions or macroscopic properties in terms of  $Q$

From thermodynamics..... Helmholtz free energy  $A = E - TS = E - kT \ln Q - T \frac{E}{T}$

$$\boxed{A = -kT \ln Q}$$

Monumentally important result!

From thermodynamics,  $dA = -pdV - SdT + \mu dN$  (single component system)

pressure  $p = -\left(\frac{\partial A}{\partial V}\right)_{T,N}$   $p = kT \left(\frac{\partial \ln Q}{\partial V}\right)_{T,N}$

chemical potential  $\mu = \left(\frac{\partial A}{\partial N}\right)_{T,V}$   $\mu = -kT \left(\frac{\partial \ln Q}{\partial N}\right)_{T,V}$

$$\left. \begin{array}{l} H = E + pV \\ G = A + pV \end{array} \right\} \begin{array}{l} \text{Homework:} \\ \text{Write in terms of } Q \end{array}$$

Now we have a framework for relating microscopic properties, as given by  $Q$ , to macroscopic properties.

Note that  $Q(E_j)$  or  $P_j(E_j)$  tells us the distribution of assembly states in the ensemble. Only the energy of an assembly state determines its probability.  $Q$  and  $P_j$  don't depend on any other properties of the states.

### Alternate form for the probabilities

Sometimes, more useful than  $P_j$  - probability of distinguishable state  $j$  - is  $P(E)$ , probability of finding an assembly with energy  $E$ .

Recall  $Q(N, V, T) = \sum_j e^{-E_j/kT} = \dots + e^{-E_\alpha/kT} + e^{-E_\beta/kT} + \dots$

But many distinguishable states are degenerate, e.g.  $E_\alpha = E_\beta = E_\gamma \equiv E$

then  $Q(N, V, T) = \dots + 3e^{-E/kT} + \dots = \sum_E \Omega(N, V, E) e^{-E/kT}$

$\Omega(N, V, E) \equiv$  degeneracy  $\equiv$  # distinguishable assembly states with energy  $E$

$$Q(N, V, T) = \sum_j e^{-E_j/kT} = \sum_E \Omega(N, V, E) e^{-E/kT}$$

sum over  
assembly states

sum over assembly  
energy levels

$$P(E) = \sum_{j \in \{E_j=E\}} P_j = \sum_{j \in \{E_j=E\}} e^{-E_j/kT} / Q(N, V, T)$$

sum over those assembly states with  $E_j = E$

$$P(E) = \frac{\Omega(N, V, E) e^{-E/kT}}{Q(N, V, T)}$$