

Enzyme Kinetics - Answer Sheet

7% of total						
Question	7.1	7.2	7.3	7.4	7.5	Total
Points	3	4	2	8	17	34
Score						

7.1 (3 pt)

Choose the correct alternative form(s) of the initial rate (v_0) expressions (1) and (2):

$v_0 = \frac{k_3[\mathbf{E}]_0[\mathbf{S}]_0}{[\mathbf{S}]_0 + K_M}$

$v_0 = \frac{k_3[\mathbf{E}]}{1 + K_M/[\mathbf{S}]_0}$

$v_0 = j[\mathbf{ES}]_{\max}$

$v_0 = \frac{k_3[\mathbf{E}]_0[\mathbf{ES}]_{\max}}{[\mathbf{S}]_0 + K_M}$

$v_0 = \frac{k_3[\mathbf{E}]_0}{1 + K_M/[\mathbf{S}]_0}$

$v_0 = \frac{j[\mathbf{E}]_0}{K_M + [\mathbf{S}]_0}$

7.2 (4 pt)

Choose the pair(s) of axes (y vs. x) that are expected to give a linear plot:

v_0 vs. $1/[\mathbf{S}]_0$

v_0 vs. v_0/K_M

v_0 vs. K_M/v_0

$1/v_0$ vs. $v_0/[\mathbf{S}]_0$

$1/v_0$ vs. v_0/K_M

$[\mathbf{S}]_0/v_0$ vs. $[\mathbf{S}]_0$

Theory



55TH INTERNATIONAL
CHEMISTRY OLYMPIAD
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A7-2

English (Official)

7.3 (2 pt)

Show that equation (3) takes the MM form (1) if the concentration of substrate **B** is maintained at a constant value c_0 :

Give the expression for v_{\max} in this case:

$$v_{\max} = \underline{\hspace{10cm}}$$

7.4 (8 pt)

Propose a kinetic scheme for the Enzymatic System I consistent with equation (3), showing all the intermediates and products. **Indicate** the reaction with a rate constant k .

7.5 (17 pt)

Fill in the table:

	Number of active sites	k_1	k_2	k_3	K_M
S _A					
S _B					
S _C		$1.57 \cdot 10^7 \text{ M}^{-1} \text{ s}^{-1}$			

7.5 (cont.)

Provide your calculations:

DELEGATION PRINT

7.5 (cont.)

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